Set No. 2

Max Marks: 75

III B.Tech I Semester Examinations, December 2011 NANOTECHNOLOGY

Chemical Engineering Time: 3 hours

Code No: 09A50807

Answer any FIVE Questions All Questions carry equal marks

1. Describe the mechanical, electrical and optical properties of carbon nanotubes. [15] 2. Define energy bands. Explain the energy band structures in metals, semiconductors and insulators. |15|3. Explain Chemical bath deposition technique in detail. [15]4. Discuss about the various synthesis methods for fullerenes. Give a short notes on the purification of fullerenes. [15]5. What is the future of nanomedicine? [15]6. What is sputtering? Discuss about different types of sputtering. [15]7. Distinguish between optical microscopies and electron microscopies schematically. [15]8. Explain the various characterization tools for core-shell nanoparticles. [15]

Code No: 09A50807

Set No. 4

III B.Tech I Semester Examinations, December 2011 NANOTECHNOLOGY

Chemical Engineering

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

- 1. Define conductor, semiconductor and insulator. Describe the energy band structures in metals, semiconductor and insulator. [15]
- 2. Explain the different synthesis methods for fullerenes. Explain the purification of fullerenes.
- 3. What are rotaxanes and catenanes? [15]
- 4. Explain about hydrolysis and condensation in sol-gel method. [15]
- 5. What is reactive sputtering? Explain about RF magnetron sputtering. [15]
- 6. Describe the various characterization tools for core-shell nanoparticles. [15]
- 7. Define Energy Dispersive Spectrometry (EDS) and Wavelength Dispersive Spectrometry (WDS). Explain how microanalysis of materials is done by using Scanning Electron microscope. [15]
- 8. Explain the filling of nanotubes and also the mechanism of growth of carbon nanotubes. [15]

Code No: 09A50807

Set No. 1

III B.Tech I Semester Examinations, December 2011 NANOTECHNOLOGY

Chemical Engineering

Time: 3 hours Max Marks: 75

Answer any FIVE Questions All Questions carry equal marks

1.	Discuss in detail about Photoelectron Spectroscopy.	[15]
2.	Describe the possible applications of nanotechnology.	[15]
3.	Explain the synthesis and purification methods for carbon nano tubes.	[15]
4.	Discuss about the optical properties of fullerenes.	[15]
5.	Write an essay on e-beam evaporation. Distinguish between thermal evaporand e-beam evaporation.	raton [15]
6.	What are the systems used and how nanoparticles are useful in the field of narology?	nobi- [15]
7.	Explain the biological applications of core-shell nanoparticles schematically.	[15]
8.	Describe briefly the deposition of some oxide-based nanocrystalline thin film Sol-Gel dip coating technique.	n via [15]

Code No: 09A50807

Set No. 3

Max Marks: 75

Time: 3 hours

Answer any FIVE Questions All Questions carry equal marks

All Questions carry equal marks $ {}^{\star\star\star\star\star}$		
1. Discuss about targeted drug delivery using nanoparticles.	[15]	
2. Explain the following:		
(a) Metallo fullerenes(b) Doped fullerenes	[4 =1	
(c) Endohedral fullerenes.	[15]	
3. Write an essay on the density of states at low- dimensional structures.	[15]	
4. Explain thermal evaporation in detail.	[15]	
5. What are the synthesis methods available for the preparation of carbon nano tubes Explain. [15]		
6. What are functionalized metal nanoparticles? Describe the advantages alized metal nanoparticles.	of function- [15]	
7. Describe various stages in Pulsed Laser Deposition process.	[15]	
8. What is the basic principle in Scannaing Electron Microscope? How different from Optical Microscopy- Explain.	w are they [15]	